



# CO<sub>2</sub> and N<sub>2</sub>O Emissions after Digestate Application Responded Differently to Topsoil Dilution and Soil Erosion State

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## Abstract

The application of digestates supports soil fertility by restoring soil organic matter (SOM) and supplying nitrogen (N). However, their application can increase greenhouse gas (GHG) emissions from agriculture. Targeted digestate application to soils where erosion has mixed subsoil into topsoil may reduce emissions, as the lower SOM saturation in this diluted topsoil could enhance stabilization of organic inputs. This study investigates whether topsoil dilution through erosion can reduce carbon dioxide (CO<sub>2</sub>) and nitrous oxide (N<sub>2</sub>O) emissions following digestate application. We conducted an incubation experiment simulating erosion-induced topsoil dilution. Three different soils from Uckermark region, Germany—non-eroded (LL), moderately eroded (eLL), and strongly eroded (RZ)—were incubated for 26 days in an automated gas exchange system. Topsoil was diluted with 20% subsoil, and digestate applied as organic fertilizer. CO<sub>2</sub> and N<sub>2</sub>O emissions were measured; undiluted, unfertilized soils served as controls. Digestate increased CO<sub>2</sub> and significantly raised N<sub>2</sub>O emissions in all soils. Topsoil dilution reduced CO<sub>2</sub> emissions in LL and showed similar trends in eLL and RZ, though the effect weakened with erosion severity, likely due to existing C-undersaturation in these soils. N<sub>2</sub>O responses varied: emissions decreased in eLL (high clay and reactive mineral content), possibly due to enhanced N stabilization, but increased in RZ (calcareous, high-pH soil likely promoting nitrification) and slightly in LL, possibly due to lowered carbon-to-nitrogen (C:N) ratio. Topsoil dilution can mitigate digestate-induced CO<sub>2</sub> but may elevate N<sub>2</sub>O emissions depending on soil properties. Therefore, site-specific management is key to lowering GHGs in erosion-prone soils.

**Keywords** Nitrous oxide · Carbon dioxide · Erosion · Topsoil dilution · Digestate fertilization

## 1 Introduction

The use of organic fertilizers is an essential farming practice for maintaining the fertility of arable soils. It facilitates the restoration and maintenance of soil organic matter (SOM) and the supply of plant nutrients, such as nitrogen (N) (Diacono and Montemurro 2011; Larney and Angers 2012). However, the risk of environmental pollution is higher with organic N fertilizers than with synthetic ones in

case of improper use, such as application to the soil surface or outside the growing season (Rashmi et al. 2020; Zhang et al. 2012). What is currently important is its contribution to the increased role of arable land as a source of climate-relevant greenhouse gases (GHG), including nitrous oxide (N<sub>2</sub>O) and carbon dioxide (CO<sub>2</sub>) (Loubet et al. 2011; Möller 2015; Smith et al. 2008). This seems to be particularly true for digestate, a nutrient-rich byproduct produced in ever-increasing quantities worldwide when biogas is extracted from plant waste or organic fertilizer (Möller et al. 2015; O'Connor et al. 2022; Revuelta-Aramburu et al. 2025). Compared to other organic fertilizers, such as slurry, digestate contains higher levels of volatile ammonium nitrogen (NH<sub>4</sub><sup>+</sup>-N), has a higher pH value, and contains a lower proportion of organic carbon (OC) compounds (Jäger et al. 2013; Möller et al. 2015; Nkoa 2014; Zhang et al. 2024). There are various indications that this is associated with a higher potential for ammonia (NH<sub>3</sub>) volatilization and

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$\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions after application to fields than with other organic or synthetic N fertilizers (Johansen et al. 2013; Rosace et al. 2020). In some cases, extremely high  $\text{N}_2\text{O}$  emissions following digestate application have been observed (Dietrich et al. 2020; Ma et al. 2011; Wang et al. 2014).

An option for mitigating the environmental and climate impact of digestate and other organic fertilizers is their preferential application to soils where carbon (C) and nutrient-poor subsoil has recently been mixed into the topsoil. This topsoil dilution is very often the result of soil erosion (Gregorich et al. 1998; Pehlivan et al. 2025; Pennock 1998; Vaidya et al. 2024; Zentgraf et al. 2024) and typically occurs on the slopes of hilly farmland in the aftermath of erosion events. In particular, plowing after an erosion event mixes the subsoil into the reduced topsoil layer (Gerke and Hierold 2012; Sommer et al. 2008). Consequently, the amount of nutrients and organic soil carbon (SOC) also decreases significantly. This usually leads to soil C undersaturation (Hassink 1997; Stewart et al. 2007; Xiao et al. 2018). This again increases the soil's ability to stabilize C by accumulating OC in the form of microbial necromass or mineral-associated organic matter (MAOM) (Doetterl et al. 2025; Georgiou et al. 2025). In other words, erosion-induced topsoil dilution induces a temporary increase in the accumulation of assimilate C in the diluted topsoil until the original SOC content and stock is restored, a phenomenon also referred to as “dynamic replacement” (Dialynas et al. 2017; Doetterl et al. 2016; Harden et al. 1999; Larney et al. 2016; Remus et al. 2018). While many details are still unclear, this process is apparently linked to reduced mineralization of OC and thus reduced  $\text{CO}_2$  emissions, potentially resulting in a net  $\text{CO}_2$ -C sink on eroded soils (Doetterl et al. 2016; Hoffmann et al. 2017, 2018; Lal 2019; Vaidya et al. 2021). However, it remains unclear whether this  $\text{CO}_2$ -C sink function, which is associated with reduced  $\text{CO}_2$  emissions from eroded soils, also operates in the context of digestate application (Levavasseur et al. 2022; Reuland et al. 2022).

Because of the close link between the C and N cycles, it is reasonable to assume that erosion-induced topsoil dilution leads to N undersaturation as well. This should result in increased temporary immobilization or stabilization of N in the form of the accumulation of organic N compounds in the soil (Bingham and Cotrufo 2016; Castellano et al. 2012; Knicker 2011; van Groenigen et al. 2015). So far, there is very little concrete information if this really takes place. However, a model experiment by Schoof et al. (2025) have shown that erosion-induced dilution of topsoil can lead to reduced gross N mineralization and  $\text{N}_2\text{O}$  emissions. Yet, it remains unclear if the  $\text{NH}_4^+$ -N applied with digestate behaves the same way in the soil. In other words, the

question is whether it is also subject to increased immobilization, stabilization, and reduced  $\text{N}_2\text{O}$  emission.

On eroded slopes, however, there is a small-scale sequence of soils with different erosion states. The closer the original C-horizon (parent material) is to the soil surface because of soil erosion, the greater the erosion state. Depending on erosion state, the topsoil is diluted with subsoil material of very different chemical-physical composition (Doetterl et al. 2025; Gerke and Hierold 2012; Gerke et al. 2016; Sommer et al. 2008, 2016). There is now some evidence that the erosion state that a soil type exhibits can have a strong impact on  $\text{CO}_2$  and  $\text{N}_2\text{O}$  emissions (Holz and Augustin 2021; Vaidya et al. 2023, 2024). In addition, high small-scale variability in  $\text{N}_2\text{O}$  emissions at erosion sites has been linked to the distribution of different erosion states in the soil (Gu et al. 2011; Vaidya et al. 2023; Vilain et al. 2010).

To address these knowledge gaps, this study focuses on understanding the role of erosion status and digestate fertilization in shaping GHG dynamics and nutrient retention under controlled conditions. We compared three soil types representing increasing erosion states: non-eroded Luvisol (LL), eroded Luvisol (eLL), and strongly eroded Calcaric Regosol (RZ), from the hummocky, north-eastern German young moraine landscape characterised by varying degrees of erosion over short distances. To assess the effects of erosion-induced topsoil dilution on  $\text{CO}_2$  and  $\text{N}_2\text{O}$  emissions, we incubated the soils under controlled conditions, applying digestate fertilizer from biogas production and diluting the topsoil with subsoil. To this end, we hypothesized that (i) erosion induced topsoil dilution and reduction of the C and N contents reduce  $\text{CO}_2$  and  $\text{N}_2\text{O}$  emissions from digestate application and that (ii) the soil properties dominated by the erosion state of a soil modify this effect.

## 2 Materials and Methods

### 2.1 Studied Soil Types

The investigations were carried out on substrates of a non-eroded Albic Luvisol (Cutanic; LL), eroded Albic Luvisol (Cutanic; eLL), and strongly eroded Calcaric Regosol (RZ) from the erosion-affected “CarboZALF-D” experimental area in the Uckermark region, where tillage erosion is the dominant erosion process. These soils were chosen because they are characteristic of large areas in the northern hemisphere, particularly post-glacial regions (Gerke and Hierold 2012; Rieckh et al. 2012). The Uckermark region is located in the hummocky arable young moraine landscape of north-east Germany (53° 23' N, 13° 47' E; ~50–60 m a.s.l). The site experiences a long-term average annual air temperature

of 9.1 °C and precipitation of 505 mm as recorded by ZALF research station in Dedelow, between the years 2005 to 2020.

The soils in this area are derived from a variety of sandy to marly glacial and glaciofluvial deposits, shaped significantly by soil erosion processes. Consequently, they exhibit distinct soil profiles that reflect different erosion states (Sommer et al. 2008, 2016; Vaidya et al. 2023). In this hummocky moraine landscape, the soil types also represent these erosion states, which are defined by the depth to the parent material (dense glacial till; C horizon; Wilken et al. 2020). The non-eroded Albic Luvisol (LL) retains a clay-eluviated E horizon beneath the Ap, and admixing this E material into the Ap mainly causes SOC dilution but does not substantially alter texture or depth to the C horizon. In the eroded Albic Luvisol (eLL), the E horizon has been removed and the Bt horizon directly underlies the Ap; admixing this Bt material likewise affects SOC but only marginally other soil properties. The strongly eroded Calcaric Regosol (RZ) represents the most advanced erosion stage, where plowing already reaches into the calcareous glacial till, and admixture introduces calcium carbonate (CaCO<sub>3</sub>)-rich material into the Ap. Thus, the selected soils correspond to realistic erosion states and distinct subsoil horizons that would be mixed into the topsoil under agricultural tillage. Table S1 illustrates notable differences in soil depth, pH values, and texture between the topsoil and subsoil horizons, which are a result of the impact of erosional processes on the development of the soil profile.

## 2.2 Incubation Experiment Setup and Treatment Preparation

For each soil type, four treatments were prepared: (1) undiluted and unfertilized topsoil (Ap), serving as a control, (2) diluted and unfertilized Ap with 20% material from the respective subsoil horizon (+), (3) undiluted, digestate fertilized Ap material (ORG), and (4) diluted, digestate fertilized Ap with 20% material from the respective subsoil horizon (+ORG). The chosen 20% subsoil admixture reflects the proportion typically incorporated into topsoils during fractional deep tillage or field manipulation experiments in the “CarboZALF-D” experimental area (Deumlich et al. 2017; Sommer et al. 2016; Vaidya et al. 2024; Zentgraf et al. 2024). These treatment abbreviations are used consistently throughout the manuscript. Preparation was conducted as follows: bulk samples from Ap and subsoil horizons from non-eroded Albic Luvisol (LL), eroded Albic Luvisol (eLL), and strongly eroded Calcaric Regosol (RZ) were collected. The samples were air dried and sieved with a 2 mm mesh.

The target aims for preparation were 55% water-filled pore space (WFPS) and a bulk density of 1.4 g cm<sup>-3</sup>, consistent across all treatments. For the control (undiluted, unfertilized) soils (LL, eLL, RZ), the moisture content of the Ap material was determined, and wet masses were calculated to obtain 4550 g of dry soil. Water (864.4 mL) was added, and the mixture was thoroughly mixed. A total of 415 g of wet soil was placed into each 250 cm<sup>3</sup> piercing cylinder (13 cylinders total) and compacted to the target bulk density. For the +treatment (diluted, unfertilized), the moisture content of both Ap and the respective subsoil horizons was determined. Wet masses were calculated to obtain 3540 g dry Ap and 910 g subsoil material. The appropriate amount of water (864.4 mL) was added, and the soil was thoroughly mixed before being placed into the piercing cylinders and compacted to the target bulk density.

The ORG treatment (undiluted, digestate fertilized) received biogas digestate from the Möllhoff farm (Uckermark): 170 kg NH<sub>4</sub><sup>+</sup>-N per ha, i.e. 20 mg NH<sub>4</sub><sup>+</sup>-N per 100 g of dry soil. The characteristics of the applied digestate are shown in Table S2. For this treatment, 4550 g of dry Ap material was mixed with 309.4 mL of digestate and the calculated amount of water (excluding the digestate volume) to reach the target water volume of 864.4 mL. The mixture was then transferred into a 250 cm<sup>3</sup> piercing cylinder (13 cylinders in total) and compacted lightly. For +ORG treatment (diluted, digestate fertilized), 3540 g of dry Ap and 910 g of subsoil material were mixed with 309.4 mL of digestate and the appropriate amount of water (excluding the digestate volume) to reach 864.4 mL. The mixture was then transferred into a 250 cm<sup>3</sup> piercing cylinder (13 cylinders in total) and compacted lightly. Additional cylinders were used to assess changes in the soil chemical properties, with 100 cm<sup>3</sup> of wet soil (with or without the admixture and fertilizer, depending on the treatment) repacked into each cylinder at the target bulk density.

## 2.3 Incubation System and Calculations

To determine the CO<sub>2</sub> and N<sub>2</sub>O emissions, the soil samples were transferred to an incubation system presented in detail in Rillig et al. (2021) (Figure S1). According to Livingston and Hutchinson (1995), the system can be described as a flow-through steady-state system, comprised of 16 sealed, cylindrical incubation vessels, each constructed from commercially available KG DN sewer pipes and accessories (Marley, Germany), with a diameter and height of 13 cm. The incubation was conducted under controlled conditions to promote microbial activity and potential N<sub>2</sub>O formation. A constant temperature of 20 °C was maintained within the incubation vessels in a controlled climate box, and moisture was adjusted to an optimal 55% WFPS to ensure favorable

conditions for nitrification and denitrification processes (Bateman and Baggs 2005; Chen et al. 2008; Dobbie and Smith 2003). Ambient air, flowing at a rate of 32 mL min<sup>-1</sup> (determined by flow controller; MKS Instruments GE50A, Andover, MA, USA), was continuously circulated through the headspace of the incubation vessels. A multiplexer was used to connect the different vessels each for 7 min after each other to a cavity ring-down spectroscopy (CRDS) Picarro G2508 gas concentration analyser (PICARRO, INC., Santa Clara, USA). Ambient air circulating directly from the pressure vessel to the gas analyser at the same flow rate, measured once after the 16 vessels served as a reference. Air circulation between the incubation unit headspace and the CRDS analyser was maintained at 250 mL min<sup>-1</sup> using a low-leak diaphragm pump (A0702, Picarro, Santa Clara, CA, USA). To prevent desiccation of the soil cores, incoming air was pre-saturated to 100% relative humidity before entering the incubation vessels using water traps. The soils were incubated under these conditions for 26 days, during which CO<sub>2</sub> and N<sub>2</sub>O emissions were measured.

N<sub>2</sub>O and CO<sub>2</sub> fluxes were calculated as follows;

$$F = \frac{(M * \rho * V * (\Delta c))}{(C * R * t * T)} \quad (1)$$

Where F is the flux rate (µg CO<sub>2</sub>-C or N<sub>2</sub>O-N core<sup>-1</sup> h<sup>-1</sup>), M is the molar mass of CO<sub>2</sub> or N<sub>2</sub>O, respectively (µg mol<sup>-1</sup>), ρ the atmospheric pressure (Pa), V is the air flow rate into the headspace and the channels (m<sup>3</sup> h<sup>-1</sup>), Δc is the difference of gas concentrations [mol] between outlet of a specific vessel and the control channel, C is the core, R the gas constant (m<sup>3</sup> Pa K<sup>-1</sup> mol<sup>-1</sup>), t is the time over which the concentration change was observed, and T the incubation temperature (K). Per-core fluxes (F<sub>core</sub>; µg CO<sub>2</sub>-C or N<sub>2</sub>O-N core<sup>-1</sup> h<sup>-1</sup>) were converted to areal fluxes (F<sub>area</sub>; µg m<sup>-2</sup> h<sup>-1</sup>) by dividing F<sub>core</sub> by the soil surface area of each incubation core (A<sub>core</sub>; m<sup>2</sup>). Core area (A<sub>core</sub>) was calculated as A<sub>core</sub> = π(d/2)<sup>2</sup>, where d is the internal diameter. Because A<sub>core</sub> is in cm<sup>2</sup>, it was converted to m<sup>2</sup> by multiplying by 10<sup>-4</sup>. An adapted version of an R script developed by Rillig et al. (2021) was used to calculate the current gas flux rates and CO<sub>2</sub>-C and N<sub>2</sub>O-N gas losses accumulated over time intervals. Additionally, the percentage of fertilizer induced CO<sub>2</sub>-C and N<sub>2</sub>O-N losses were calculated by subtracting the cumulative CO<sub>2</sub> and N<sub>2</sub>O emissions of unfertilized treatments from the cumulative CO<sub>2</sub> and N<sub>2</sub>O emissions of fertilized treatments. Thereafter, the percentage (%) of fertilizer induced CO<sub>2</sub>-C and N<sub>2</sub>O-N losses were calculated by multiplying the CO<sub>2</sub>-C and N<sub>2</sub>O-N losses by 100 and dividing by the amount of C (973.6 mg) and N (70 mg) that was present in the applied digestate fertilizer (Table 2). Thus, we were able

to calculate the fertilizer induced CO<sub>2</sub>-C and N<sub>2</sub>O-N emissions as fractions of added C and N through fertilization.

## 2.4 Soil Analyses

Soil samples were collected from the additional 100 cm<sup>3</sup> cylinders at selected times during the 26-day incubation period: t0 (start of the experiment), t1 and t2 (during N<sub>2</sub>O peak) and t3 (end of the experiment).

At the time points: t0, t1, t2 and t3 the following soil properties were analysed by the central laboratory at ZALF: ammonium-N (NH<sub>4</sub><sup>+</sup>-N), nitrate-N (NO<sub>3</sub><sup>-</sup>-N), pH value, cold water soluble C (CWSC) and cold water soluble N (CWSN). The determination of NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N was done according to ISO 14,256 using a CFA-SAN (Skalar analytic GmbH, Breda, Netherlands). The pH values were determined based on ISO 10,390, using a 855 Robotic Titrosampler (Deutsche Metrohm GmbH & Co. KG, Germany) in a 1:5 (volume fraction) suspension of soil in 0,01 mol/L calcium chloride solution (pH in CaCl<sub>2</sub>). CWSC and CWSN values that reflect the labile C and N pools in the soil were determined with a TOC-V CPH analyser (Shimadzu Deutschland GmbH, Germany). At the end of the experiment (t3), the total C (TC) and total organic carbon (TOC) were analyzed according to ISO 10,694 using a multiphase determinator (RC 612, Leco Instruments GmbH, Mönchengladbach, Germany). Additionally, total N (TN) was analysed at by a CNS928-MLC elemental analyser (LECO Ltd., Germany) according to ISO 13,878. Pedogenic iron (Fe) and aluminum (Al) oxides were characterized by sodium dithionite extraction (Fe<sub>dith</sub>, Al<sub>dith</sub>) and acid ammonium oxalate extraction (Fe<sub>ox</sub>, Al<sub>ox</sub>) following Schlichting et al. (1995). The Fe and Al concentrations in the extracts were determined by ICP-OES (ThermoFisher SCIENTIFIC GmbH, Germany).

## 2.5 Statistics

To evaluate the effects of soil type (erosion state), topsoil dilution, and digestate fertilisation, we performed a three-way analyses of variance (ANOVA) on cumulative CO<sub>2</sub>-C and N<sub>2</sub>O-N emissions, fertilizer-induced C and N losses, and final soil chemical properties (TC, TOC, TN, Al<sub>ox</sub>, Al<sub>dith</sub>, Fe<sub>ox</sub>, and Fe<sub>dith</sub>). These models considered the three factors as fixed effects and their interactions. Where ANOVA assumptions were met, post hoc pairwise comparisons were conducted using Tukey's Honest Significant Difference (HSD) test (α=0.05). The normality of residuals was evaluated using the Kolmogorov-Smirnov test, and homogeneity of variances was assessed using Levene's test. Residual plots were also visually inspected. In cases where normality assumptions were violated and could not

be corrected by transformation (e.g., for cumulative N<sub>2</sub>O emissions) non-parametric Kruskal Wallis tests were used, followed by Dunn's test for pairwise comparisons with Bonferroni correction.

For CWSC (Table S4) CWSN (Table S5), NH<sub>4</sub><sup>+</sup>-N (Table S6), and NO<sub>3</sub><sup>-</sup>-N concentrations (Table S7), linear mixed-effects models (LMMs) were used to assess the effects of fertilization, topsoil dilution, soil type, time, and their interactions. Time point (t0–t3) was modelled as a fixed effect, while replicate ID was included as a random effect to account for repeated measures. Post hoc pairwise comparisons were conducted on estimated marginal means (EMMs) using Tukey adjustment for multiple testing.

Dynamic CO<sub>2</sub> (Table S8) and N<sub>2</sub>O fluxes (Table S9) over time were analysed separately using LMMs, with fixed effects for treatment, day of measurement (DOM), and their interaction (Treatment x DOM), and random intercepts for experimental repetitions. Separate models were fitted for different treatment groups (e.g., undiluted soils, diluted soils, fertilized soils, and their combinations). EMMs and pairwise Tukey-adjusted contrasts were extracted to identify significant differences between treatments over time.

All data are presented as means ± standard deviation (SD). Compact letter displays were used to denote significant differences between treatments. All statistical analyses were conducted in R version 4.2.1 (R Core Team, 2022).

### 3 Results

#### 3.1 Erosion-Induced Topsoil Dilution Effects on Stable and Dynamic Soil Properties

TC (%) decreased in the LL and eLL soils as a result of topsoil dilution, regardless of fertilizer addition. However, in the RZ soil, topsoil dilution did not significantly affect the unfertilized treatment, while there was a slight increase in the fertilized treatment compared to the other soil types. In addition, the consistently clear difference between TC (%) and TOC (%) contents in the RZ indicates a significantly higher proportion of carbonate C than in the other soils. TOC (%) decreased across all soil types, in both the fertilized and unfertilized treatments. TN (%) decreased slightly as a result of topsoil dilution only in the fertilized treatment of the LL soil, while both fertilized and unfertilized treatments of the eLL and RZ soils showed a slight decrease (Table 1). Additionally, topsoil dilution slightly increased the clay bound Al and Fe oxide (Al<sub>ox</sub>, Al<sub>dith</sub>, Fe<sub>ox</sub>, Fe<sub>dith</sub>) contents (mg kg<sup>-1</sup>) in the eLL soil, while only minor changes were observed in LL and RZ. Fertilization caused small increases in Al<sub>ox</sub> in LL and eLL and in Fe<sub>ox</sub> contents in eLL, whereas the other

**Table 1** Soil chemical properties ((%) soil dry matter (DM)), including total soil carbon (TC), total soil organic carbon (TOC) and total nitrogen (TN) for different soil types and erosion states: non-eroded albic luvisol (LL), eroded albic luvisol (eLL), and strongly eroded calcareous regosol (RZ), at the end of the experiment (t3), after 26 days. Treatments include control (undiluted, unfertilized), + (diluted, unfertilized), ORG (undiluted, digestate fertilized), and +ORG (diluted, digestate fertilized). The displayed values represent the mean of the replicate treatments ± standard deviation (*n*=4). Letters represent statistically significant variations (*p*-value, 0.05 threshold) among the treatments, based on the three-way ANOVA

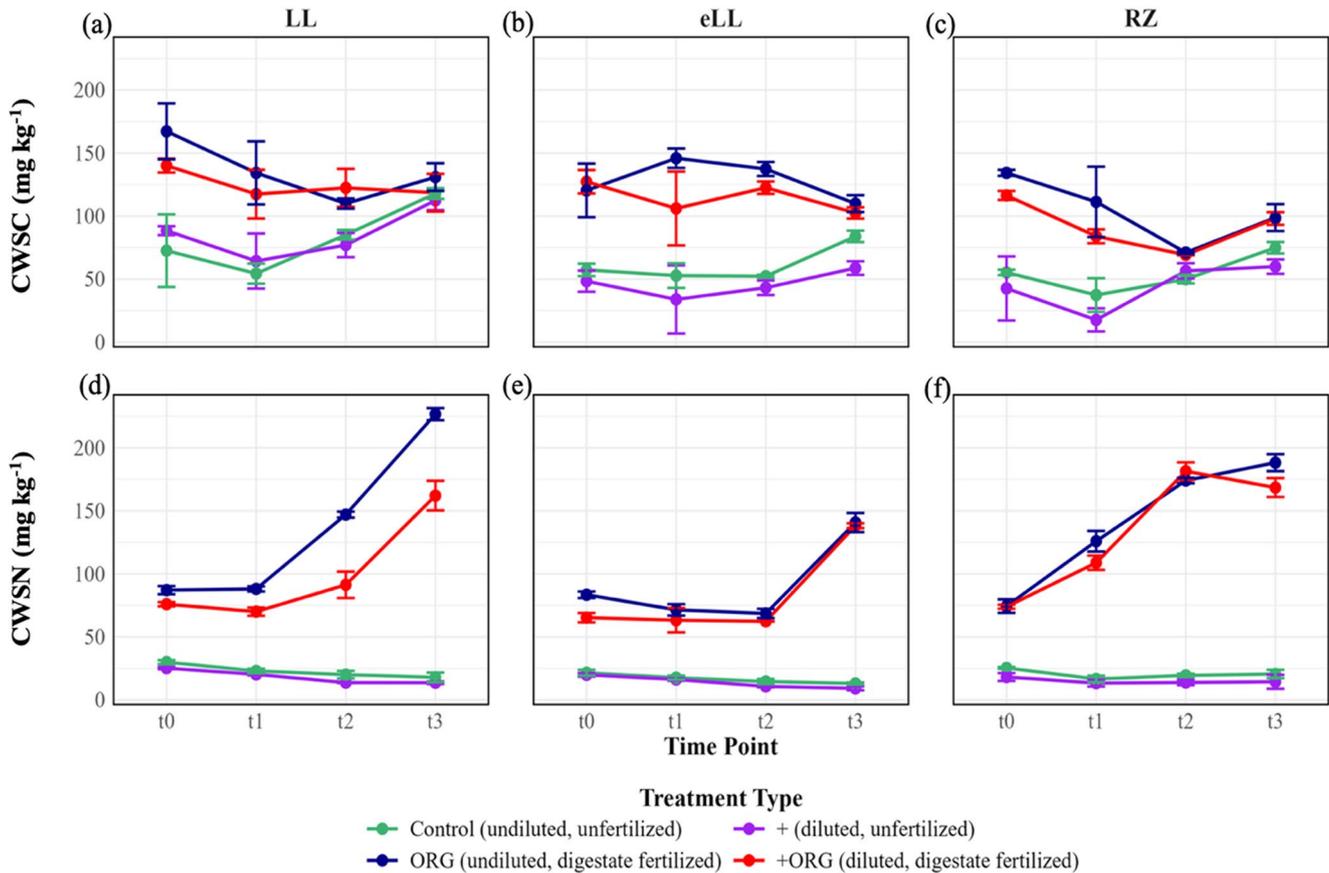
Soil type	TC (% soil DM)	TOC (% soil DM)	TN (% soil DM)
LL	0.61 <sup>h</sup> ± 0.01	0.59 <sup>e</sup> ± 0.01	0.06 <sup>e</sup> ± 0.01
+LL	0.54 <sup>i</sup> ± 0.02	0.52 <sup>f</sup> ± 0.02	0.05 <sup>e</sup> ± 0.01
LL ORG	0.84 <sup>ef</sup> ± 0.04	0.82 <sup>c</sup> ± 0.04	0.10 <sup>ab</sup> ± 0.01
+LL ORG	0.73 <sup>g</sup> ± 0.02	0.71 <sup>d</sup> ± 0.02	0.09 <sup>bc</sup> ± 0.00
eLL	0.79 <sup>fg</sup> ± 0.03	0.73 <sup>d</sup> ± 0.03	0.08 <sup>cd</sup> ± 0.01
+eLL	0.66 <sup>h</sup> ± 0.02	0.61 <sup>e</sup> ± 0.02	0.06 <sup>de</sup> ± 0.00
eLL ORG	1.01 <sup>d</sup> ± 0.01	0.95 <sup>a</sup> ± 0.01	0.11 <sup>ab</sup> ± 0.01
+eLL ORG	0.88 <sup>e</sup> ± 0.01	0.83 <sup>bc</sup> ± 0.00	0.10 <sup>a</sup> ± 0.01
RZ	1.11 <sup>c</sup> ± 0.02	0.72 <sup>d</sup> ± 0.02	0.08 <sup>c</sup> ± 0.00
+RZ	1.15 <sup>c</sup> ± 0.04	0.56 <sup>ef</sup> ± 0.04	0.07 <sup>de</sup> ± 0.00
RZ ORG	1.26 <sup>b</sup> ± 0.02	0.89 <sup>b</sup> ± 0.01	0.11 <sup>a</sup> ± 0.01
+RZ ORG	1.37 <sup>a</sup> ± 0.05	0.79 <sup>c</sup> ± 0.02	0.10 <sup>ab</sup> ± 0.01

oxide forms remained largely unchanged. In the fertilized treatments, topsoil dilution led to a slight increase in oxide contents in eLL, but small decreases or no clear changes in LL and RZ (Table S3).

CWSC concentrations (mg kg<sup>-1</sup> soil) were consistently higher in digestate-fertilized soils across all time points. Topsoil dilution had no significant effect within soil types, except at t1, where CWSC was lower in +eLL ORG compared to eLL ORG. CWSC increased over time in fertilized soils, while unfertilized treatments showed low CWSC throughout the experiment. No differences were found between diluted and undiluted soils in unfertilized treatments (Fig. 1a–c).

CWSN concentrations (mg kg<sup>-1</sup> soil) were consistently higher in digestate-fertilized soils across all time points. Topsoil dilution generally had no significant effect within soil types, although in a few cases (e.g., +eLL ORG at t0, +LL ORG and +RZ ORG at t1 and t3), diluted treatments showed slightly lower CWSN than their undiluted counterparts. CWSN increased over time in fertilized soils, particularly between t2 and t3, while remaining low in unfertilized soils (Fig. 1d–f).

NH<sub>4</sub><sup>+</sup>-N concentrations (mg kg<sup>-1</sup> soil) were consistently higher in digestate-fertilized soils across all soil types, especially at early time points. Topsoil dilution generally had no significant effect on NH<sub>4</sub><sup>+</sup>-N concentrations within fertilized soils. However, at t1, NH<sub>4</sub><sup>+</sup>-N was significantly lower in the diluted eLL ORG treatment compared to eLL ORG, indicating a short-term dilution effect. NH<sub>4</sub><sup>+</sup>-N



**Fig. 1** Dynamics of cold water soluble carbon (CWSC; top row) and nitrogen (CWSN; bottom row) concentrations ( $\text{mg kg}^{-1}$  soil) across four time points (t0, t1, t2, t3) for different soil types and erosion states: (a, d) non-eroded Albic Luvisol (LL), (b, e) eroded Albic Luvisol (eLL), and (c, f) strongly eroded Calcaric Regosol (RZ). Treatments include control (undiluted, unfertilized; green), + (diluted, unfertilized; purple), ORG (undiluted, digestate fertilized; blue), and +ORG

(diluted, digestate fertilized; red). Statistical analysis was conducted using linear mixed-effects models (LMMs), with replicate ID as a random effect to account for repeated measurements across time. Fixed effects included fertilization, topsoil dilution, soil type, time, and their interactions. Data represent means of independent replicates ( $n=3$  for t0–t2;  $n=4$  for t3). Error bars indicate  $\pm 1$  standard deviation

levels declined rapidly over time in all fertilized treatments, approaching zero by the end of the incubation. In unfertilized soils,  $\text{NH}_4^+$ -N remained low throughout the experiment (Fig. 2a–c).

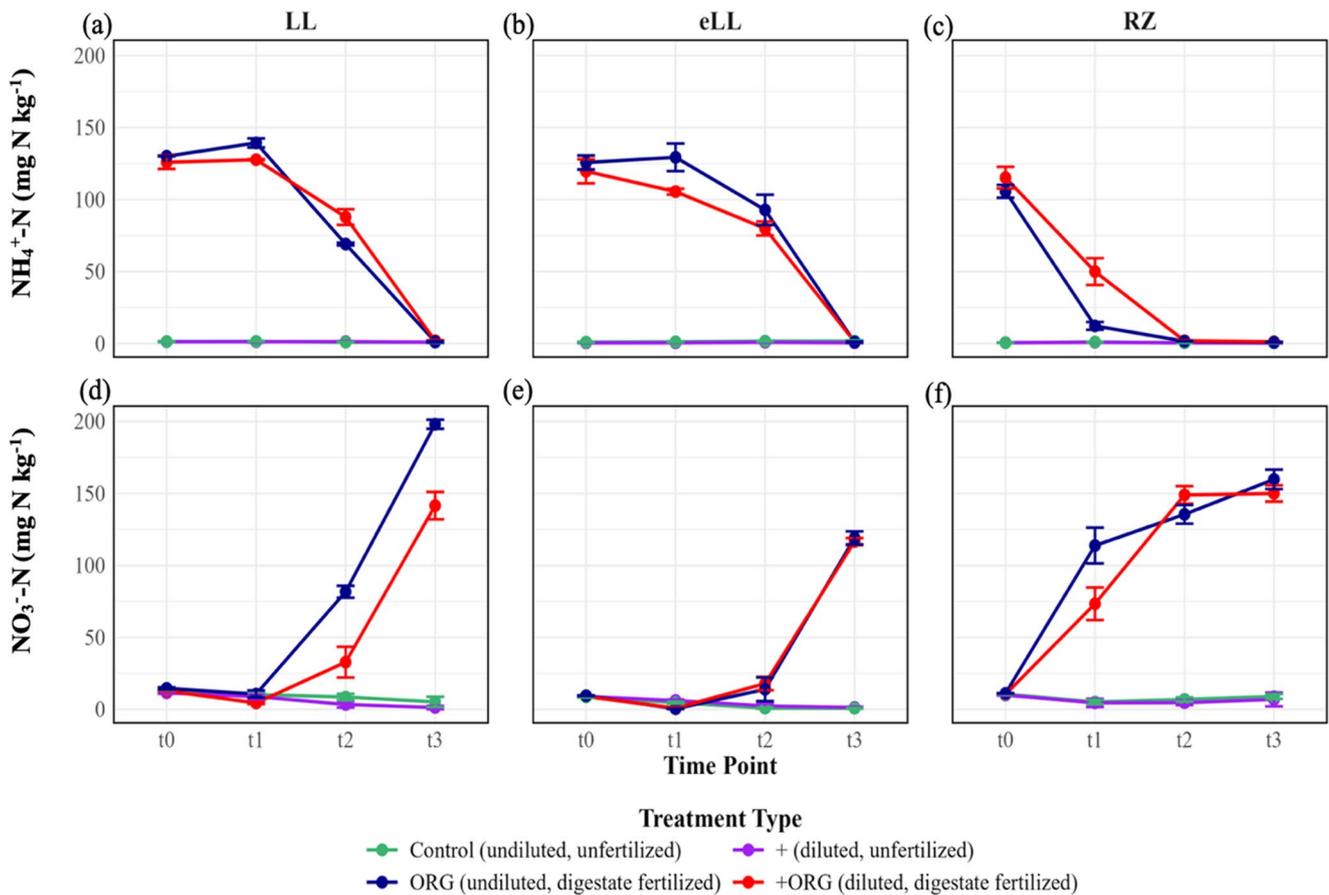
$\text{NO}_3^-$ -N concentrations ( $\text{mg kg}^{-1}$  soil) increased substantially over time in digestate-fertilized soils, reaching significantly higher levels at t3. Topsoil dilution had no consistent effect overall, though at t1, the undiluted LL ORG treatment had significantly higher  $\text{NO}_3^-$ -N than the diluted LL treatment. In unfertilized treatments,  $\text{NO}_3^-$ -N levels remained low and showed only minor variation throughout the incubation. These trends align with significant treatment-by-time interactions detected in the statistical analysis (Fig. 2d–f).

### 3.2 $\text{CO}_2$ and $\text{N}_2\text{O}$ Flux Dynamics

$\text{CO}_2$  fluxes ( $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) varied significantly across different soil types and erosion states, with the highest fluxes observed in the LL soil, followed by eLL, and the lowest in

the RZ soil (Fig. 3a). Topsoil dilution significantly reduced  $\text{CO}_2$  fluxes in the LL and RZ compared to their undiluted counterparts, with the greatest reduction observed in the LL, followed by RZ, and a slight increase observed in the eLL (Fig. 3b). Digestate application led to a significant increase in  $\text{CO}_2$  fluxes across all soils, with the strongest effect observed in the RZ, followed by LL, and a moderate increase in the eLL (Fig. 3c). In fertilized soils (LL ORG, eLL ORG, RZ ORG), topsoil dilution significantly reduced  $\text{CO}_2$  fluxes, with the most pronounced reduction in the LL ORG, followed by eLL ORG, and a smaller reduction observed in the RZ ORG treatment (Fig. 3d).

$\text{N}_2\text{O}$  fluxes ( $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) varied by soil type, with the highest emissions observed in the RZ soils, followed by LL and eLL soils (Fig. 4a). Topsoil dilution significantly reduced  $\text{N}_2\text{O}$  fluxes in all soil types compared to their undiluted (LL, eLL and RZ) counterparts (Fig. 4b). Digestate addition significantly increased  $\text{N}_2\text{O}$  emissions across all soil types, with the eLL ORG showing significantly higher



**Fig. 2** Dynamics of ammonium ( $\text{NH}_4^+\text{-N}$ ; top row) and nitrate ( $\text{NO}_3^-\text{-N}$ ; bottom row) concentrations ( $\text{mg N kg}^{-1}$  soil) across four time points (t0, t1, t2, t3) for different soil types and erosion states: (a, d) non-eroded Albic Luvisol (LL), (b, e) eroded Albic Luvisol (eLL), and (c, f) strongly eroded Calcaric Regosol (RZ). Treatments include control (undiluted, unfertilized; green), + (diluted, unfertilized; purple), ORG (undiluted, digestate fertilized; blue), and +ORG (diluted, diges-

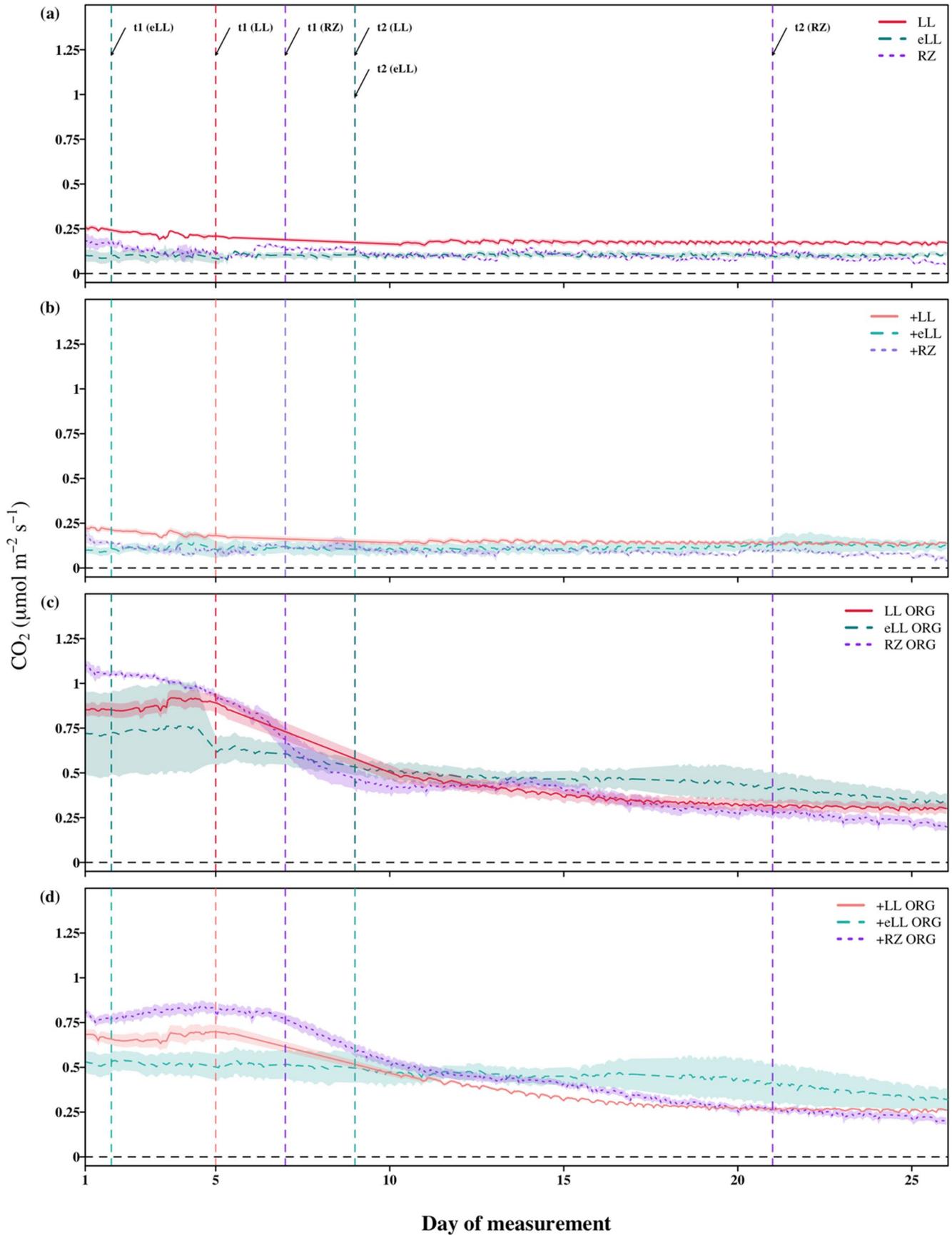
tate fertilized; red). Statistical analysis was conducted using linear mixed-effects models (LMMs), with replicate ID as a random effect to account for repeated measurements across time. Fixed effects included fertilization, topsoil dilution, soil type, time, and their interactions. Data represent means of independent replicates ( $n=3$  for t0–t2;  $n=4$  for t3). Error bars indicate  $\pm 1$  standard deviation

emissions than the LL ORG and RZ ORG (Fig. 4c) Topsoil dilution significantly increased  $\text{N}_2\text{O}$  fluxes in the +LL ORG and +RZ ORG compared to their undiluted fertilised counterparts (LL ORG and RZ ORG), whereas in +eLL ORG, fluxes were significantly lower than in eLL ORG (Fig. 4d).

### 3.3 Cumulative C and N Losses by $\text{CO}_2$ and $\text{N}_2\text{O}$

The cumulative  $\text{CO}_2$  emission ( $\text{mg CO}_2\text{-C core}^{-1}$ ) of the LL soil was the highest compared to eLL and RZ soils which had similar values among each other (Table 2). Topsoil dilution resulted in a decrease in the cumulative  $\text{CO}_2$  emissions in the +LL soil, while no differences were observed in the +eLL and +RZ soils. Digestate addition increased cumulative  $\text{CO}_2$  emissions in all soil types, while topsoil dilution decreased cumulative  $\text{CO}_2$  emissions in all these fertilized soil types. Unfertilized treatments showed the lowest  $\text{N}_2\text{O}$  emissions ( $\text{mg N}_2\text{O-N core}^{-1}$ ) and there were

no statistically significant differences between soils with and without topsoil dilution. The eLL ORG soil had the highest  $\text{N}_2\text{O}$  emissions, followed by RZ ORG and LL ORG. Topsoil dilution increased cumulative  $\text{N}_2\text{O}$  emissions in the +LL ORG and +RZ ORG, particularly in the +RZ ORG treatment. In contrast, topsoil dilution decreased  $\text{N}_2\text{O}$  emissions in the +eLL ORG treatment. Fertilizer-induced  $\text{CO}_2\text{-C}$  losses (%) were similar in the eLL ORG and RZ ORG, while LL ORG had the lowest amount. Fertilizer-induced  $\text{CO}_2\text{-C}$  losses decreased significantly with topsoil dilution in the +LL ORG and +eLL ORG soils. This trend was only slightly visible for the +RZ ORG. The eLL ORG had the highest amount of fertilizer induced  $\text{N}_2\text{O-N}$  losses (%). The LL ORG and RZ ORG treatments showed similar values, both lower than the amount of fertilizer-induced  $\text{N}_2\text{O-N}$  losses of the eLL ORG. Fertilizer  $\text{N}_2\text{O-N}$  losses were not significantly affected by topsoil dilution in the +LL ORG. However, there was a significant increase in the +RZ ORG



**Fig. 3** CO<sub>2</sub> flux dynamics ( $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) over the 26-day measurement period. Lines represent soil types and erosion states: non-eroded Albic Luvisol (LL; red, solid), eroded Albic Luvisol (eLL; turquoise, dashed), and strongly eroded Calcaric Regosol (RZ; purple, dotted). Panels show different treatments: (a) control (undiluted, unfertilized), (b) + (diluted, unfertilized), (c) ORG (undiluted, digestate fertilized), and (d) +ORG (diluted, digestate fertilized). Shaded areas represent  $\pm 1$  standard deviation (SD). Vertical dashed lines show soil sampling at t1 and t2 (during the N<sub>2</sub>O peak); colors match the respective soil types/treatments. LL and eLL overlap at t2. t0 (start) and t3 (end) coincide with the measurement period limits and are omitted for clarity. Fluxes were analyzed using separate linear mixed-effects models (LMMs) fitted for each soil group, with fixed effects for treatment (soil type, dilution, fertilization), day of measurement (DOM), and their interaction, and random intercepts for experimental repetitions. Pairwise differences between treatments were assessed using estimated marginal means (EMMs) with Tukey adjustment

and a significant decrease in +eLL ORG due to topsoil dilution (Table 2).

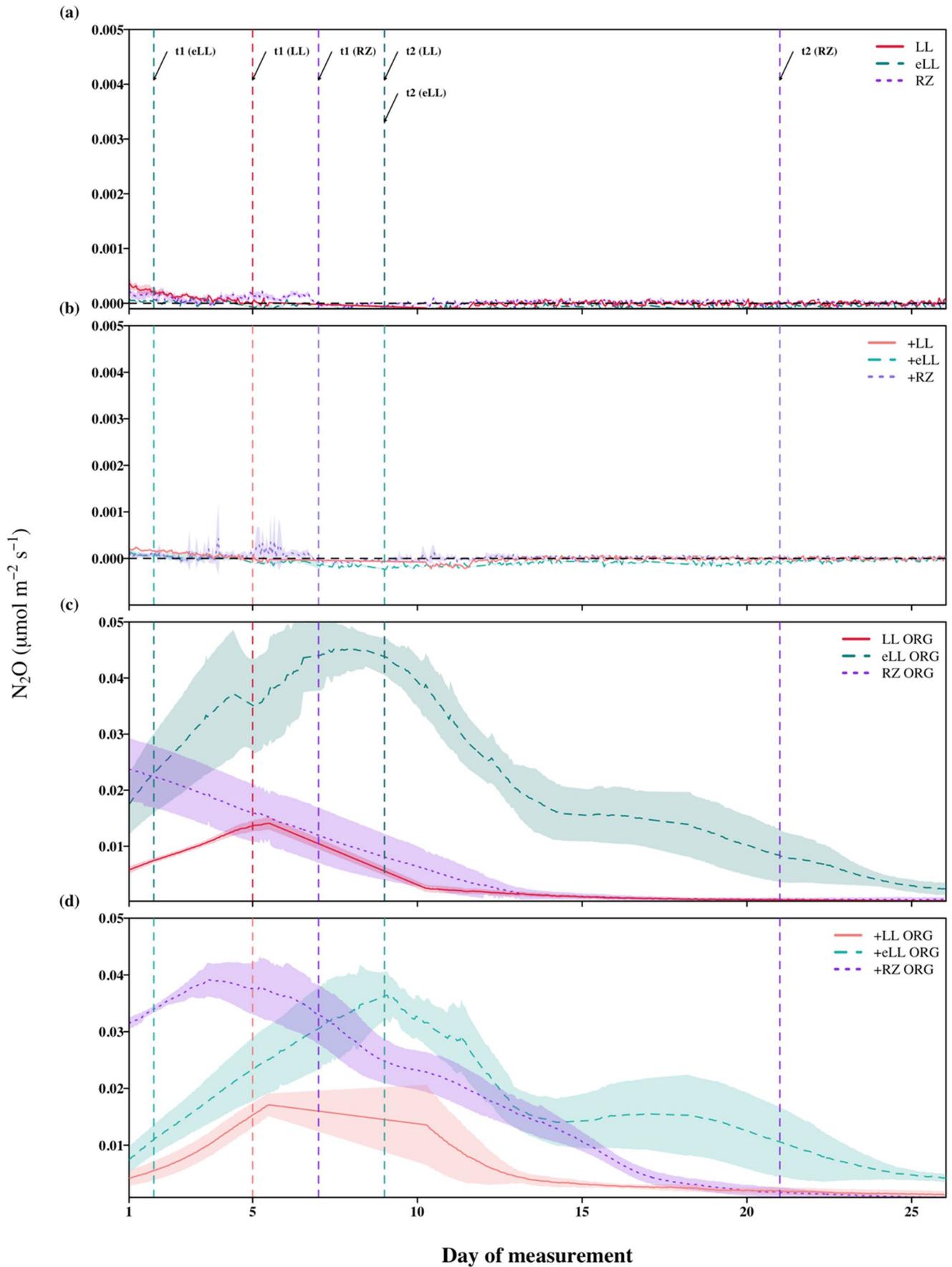
## 4 Discussion

This study investigated how erosion-induced topsoil dilution influences CO<sub>2</sub> and N<sub>2</sub>O emissions following digestate application, testing two hypotheses: (i) erosion induced topsoil dilution and reduction of the C and N contents reduce CO<sub>2</sub> and N<sub>2</sub>O emissions from digestate application and that (ii) the soil properties dominated by the erosion state of a soil modify this effect. The results confirmed that digestate application consistently increased CO<sub>2</sub>, and more strongly N<sub>2</sub>O emissions across all soils. However, topsoil dilution mitigated CO<sub>2</sub> emissions in all digestate-fertilized soils, with a significant reduction observed only in the non-eroded (LL) soil. This indicates that while the direction of the dilution effect was uniform, its strength depended on erosion severity, thereby supporting both hypotheses for CO<sub>2</sub> emissions. In contrast, N<sub>2</sub>O responses were more variable: dilution reduced emissions in the moderately eroded (eLL) soil but increased them in the strongly eroded (RZ) soil, and slightly in LL. Consequently, the first hypothesis was only partially supported for N<sub>2</sub>O, whereas the second hypothesis was clearly confirmed. These findings suggest that management strategies on eroded soils should consider the erosion state when applying digestate, as the same input may have contrasting effects on CO<sub>2</sub> and N<sub>2</sub>O emissions.

Digestate addition significantly increased CO<sub>2</sub> emissions across all soil types, as reflected in both flux measurements and cumulative emissions (Table 2; Fig. 3c). This increase corresponds with higher TOC levels (Table 1) as well as higher CWSC (Fig. 1a-c) following digestate application, suggesting that the added organic material provided readily available C sources that enhanced soil respiration and CO<sub>2</sub> release. These findings align with previous studies reporting elevated CO<sub>2</sub> emissions after digestate application due to

increased substrate availability for decomposition (Albuquerque et al. 2012; Bol et al. 2003; Fangueiro et al. 2010; Pezzolla et al. 2012; Rochette et al. 2000). Topsoil dilution, however, reduced CO<sub>2</sub> fluxes in digestate-fertilized soils across all erosion stages (Fig. 3d), but cumulative CO<sub>2</sub> emissions showed a statistically significant reduction only in the non-eroded LL soil (Table 2). This indicates that while the direction of the dilution effect was consistent, its magnitude varied, with the most pronounced mitigation occurring in LL. A similar trend was observed in unfertilized soils: fluxes decreased slightly with dilution in LL and RZ but remained unchanged or increased in eLL (Fig. 3b), again with only LL showing a significant overall reduction (Table 2). These findings suggest that the capacity of topsoil dilution to reduce CO<sub>2</sub> emissions is strongest in non-eroded soils, likely due to the decline in TOC following the incorporation of C-poor subsoil (Table 1). The decrease in SOC aligns with previous studies showing that topsoil removal reduces C and nutrient availability and alters microbial community composition (Allison and Ausden 2004; Geissen et al. 2013; Niemeyer et al. 2007; Pehlivan et al. 2025; Ruggaber et al. 2024; Zentgraf et al. 2024). However, our results indicate that the effect of topsoil dilution differs based on digestate application. Specifically, in the first 13 days of the experiment, the difference in CO<sub>2</sub> emissions between digestate-applied (Fig. 3d) and non-applied treatments (Fig. 3b) was substantial. In the remaining days, the emissions became more similar, although some differences persisted, likely due to enhanced microbial activity stimulated by digestate addition (Fig. 3d and b). Overall, this pattern suggests that topsoil dilution not only reduces SOC and CO<sub>2</sub> emissions but may also facilitate the retention of C from digestate in the soil.

One potential mechanism underlying this C retention could be the enhanced stabilization of digestate-derived OC through sorptive interactions with subsoil clay minerals. Less saturated subsoil clay minerals offer more binding sites, allowing digestate-derived C to associate with mineral surfaces, thereby reducing its bioavailability (Baldock and Skjemstad 2000; Hassink and Whitmore 1997; Kaiser and Guggenberger 2003; Kögel-Knabner et al. 2008; Sollins et al. 2009; Spielvogel et al. 2008). As mentioned, this dilution of SOC stocks likely resulted in a C undersaturation, increasing the soil's capacity for additional C input (Berhe and Kleber 2013; Doetterl et al. 2016; Hassink, 1997; Stewart et al. 2007; Xiao et al. 2018). Within this context, the concept of “dynamic replacement” might help explain the observed trend. Topsoil dilution promotes the temporary accumulation of fresh OC, in this case, OC from digestate, within the diminished C stock due to topsoil dilution, thereby reducing CO<sub>2</sub> fluxes (Dialynas et al. 2017; Doetterl et al. 2016; Harden et al. 1999; Hoffmann et al. 2017,



**Fig. 4** N<sub>2</sub>O flux dynamics ( $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) over the 26-day measurement period. Lines represent soil types and erosion states: non-eroded Albic Luvisol (LL; red, solid), eroded Albic Luvisol (eLL; turquoise, dashed), and strongly eroded Calcaric Regosol (RZ; purple, dotted). Panels show different treatments: (a) control (undiluted, unfertilized), (b) + (diluted, unfertilized), (c) ORG (undiluted, digestate fertilized), and (d) +ORG (diluted, digestate fertilized). Shaded areas represent  $\pm 1$  standard deviation (SD). Vertical dashed lines show soil sampling at t1 and t2 (during the N<sub>2</sub>O peak); colors match the respective soil types/treatments. LL and eLL overlap at t2. t0 (start) and t3 (end) coincide with the measurement period limits and are omitted for clarity. Fluxes were analysed using separate linear mixed-effects models (LMMs) fitted for each soil group, with fixed effects for treatment (soil type, dilution, fertilization), day of measurement (DOM), and their interaction, and random intercepts for experimental repetitions. Pairwise differences between treatments were assessed using estimated marginal means (EMMs) with Tukey adjustment

2018; Lal 2019), possibly leading to enhanced C sequestration. However, given the short duration of the experiment, it remains unclear whether this change in SOC represents true stabilization or merely a temporary accumulation of labile OC. Future studies should test whether such C stabilization persists under field conditions and over longer timescales to capture long-term SOC dynamics.

The erosion state further influenced the pattern of topsoil dilution effects on CO<sub>2</sub> emissions, particularly in the flux dynamics. The reduction in emissions with dilution appeared to diminish with increasing erosion severity, as seen in the flux data (Fig. 3d), possibly because highly eroded soils are already C-undersaturated, limiting the potential of dilution to further reduce emissions (Hassink, 1997; Holz and Augustin 2021; Stewart et al. 2009; West and Six 2007). Notably, the effect of dilution was more pronounced in digestate-fertilized soils, where the added organic material likely enhanced C retention and temporarily mitigated CO<sub>2</sub> emissions across all erosion stages (Levvasseur et al. 2022; Reuland et al. 2022). Nevertheless, cumulative CO<sub>2</sub> emissions (Table 2) showed a statistically significant reduction only in the non-eroded LL soil, indicating that while the direction of the dilution effect was consistent, its magnitude and duration varied with erosion severity. This pattern suggests that the potential for C retention through dilution is highest in non-eroded soils, while in more eroded soils, the existing C-undersaturation (Table 1) may limit this response over time. Overall, these findings confirm that erosion severity modulates the capacity of soils to retain C inputs from digestate and that the mitigation effect of topsoil dilution on CO<sub>2</sub> emissions is most effective under non-eroded conditions.

Beyond CO<sub>2</sub> dynamics, digestate application also had a pronounced influence on N<sub>2</sub>O emissions, although the response to topsoil dilution was more variable among soils. Digestate addition significantly increased N<sub>2</sub>O emissions across all soil types (Fig. 4c; Table 2), consistent with the observed rise in TN levels (Table 1) and CWSN

concentrations (Fig. 1d-f), indicating greater availability of mineralizable N for microbial processes. This was expected given the high NH<sub>4</sub><sup>+</sup>-N content of the digestate, which provided a readily available N source for microbial nitrification and denitrification processes (Dietrich et al. 2020; Ma et al. 2011; Wang et al. 2014). Elevated NH<sub>4</sub><sup>+</sup>-N concentrations in fertilized soils during the early stages (Fig. 2a-c) and increased NO<sub>3</sub><sup>-</sup>-N levels at later stages (Fig. 2d-f) promote microbial activity and N transformation processes, causing high N<sub>2</sub>O emissions (Batlle-Aguilar et al. 2011; Groffman et al. 2002). While topsoil dilution generally led to reduced CO<sub>2</sub> fluxes, its effect on N<sub>2</sub>O emissions was more variable, reflecting the complex and soil-dependent nature of N cycling. Specifically, N<sub>2</sub>O emissions were reduced in the moderately eroded eLL soil, increased in the strongly eroded RZ soil and showed a smaller, non-significant increase in the LL soil (Fig. 4d; Table 2). These contrasting responses suggest that the influence of topsoil dilution is not solely governed by erosion status, but also by underlying soil properties. This finding highlights key differences between C and N cycling in agroecosystems, as well as their interactions with digestate properties and soil characteristics (de la Fuente et al. 2013; Möller 2015; Reuland et al. 2022). While CO<sub>2</sub> emissions are largely controlled by TOC levels (Doetterl et al. 2016; Hoffmann et al. 2017, 2018; Lal 2019; Vaidya et al. 2021), N<sub>2</sub>O emissions are influenced by a broader set of interacting soil properties such as texture, pH, CaCO<sub>3</sub> and reactive mineral oxides, that jointly regulate N cycling (Bingham and Cotrufo 2016; Lützow et al. 2006; Yu et al. 2013). Consequently, the varying effects of dilution on these soil properties influence N transformation processes, leading to an inconsistent response in N<sub>2</sub>O emissions across different soils.

The moderately eroded eLL soil showed the most distinct response to topsoil dilution, with marked reduction in N<sub>2</sub>O emissions attributable to its clay-rich subsoil and enrichment in Fe and Al oxides. Among all soils, the eLL soil contained the highest clay content overall, and its Bt subsoil was particularly clay-rich ( $\approx 19\%$ ; Table S1). After dilution, the total clay content increased only slightly (from 14% to 15%), reflecting the minor effect of subsoil admixture rather than a statistically significant difference. However, this modest increase was accompanied by a higher abundance of reactive clay-associated minerals, as indicated by the elevated oxalate- and dithionite-extractable Al and Fe concentrations (Table S3). Previous studies on the similar soil type confirm its elevated clay fraction and oxide enrichment (Pehlivan et al. 2025; Remus et al. 2018). X-ray diffraction (XRD) analyses showed illite (30%), illite-smectite mixed-layer minerals (65%), kaolinite (5%), and traces of chlorite (Remus et al. 2018), supporting the high cation-binding capacity of the diluted eLL soil. Consistent with these properties, the

**Table 2** Cumulative (26 days) CO<sub>2</sub> (mg CO<sub>2</sub>-C core<sup>-1</sup>) and N<sub>2</sub>O (mg N<sub>2</sub>O-N core<sup>-1</sup>) emissions, fertilizer induced C and N losses as CO<sub>2</sub>-C (%) and N<sub>2</sub>O-N (%) for different soil types and erosion states: non-eroded albic luvisol (LL), eroded albic luvisol (eLL), and strongly eroded calcareous regosol (RZ). Treatments include control (undiluted, unfertilized), + (diluted, unfertilized), ORG (undiluted, digestate fertilized), and +ORG (diluted, digestate fertilized). Percentage losses of CO<sub>2</sub>-C and N<sub>2</sub>O-N were calculated relative to the amount of C (973.6 mg) and N (70 mg) applied via digestate fertilizer. Values represent means ± standard deviation (*n*=4). Statistical analysis was performed using three-way ANOVA with soil type, topsoil dilution, and fertilisation as fixed factors. Where significant effects were detected, tukey's HSD was used for post hoc comparisons. For N<sub>2</sub>O, which violated ANOVA assumptions (non-normal residuals, *p*<0.01), Kruskal-Wallis and dunn's post hoc tests (Bonferroni-corrected) were applied. Different letters indicate statistically significant differences between treatments (*p*<0.05)

Treatment	CO <sub>2</sub> (mg CO <sub>2</sub> -C core <sup>-1</sup> )	N <sub>2</sub> O (mg N <sub>2</sub> O-N core <sup>-1</sup> )	Fertilizer induced CO <sub>2</sub> -C losses (%)	Fertilizer induced N <sub>2</sub> O-N losses (%)
LL	5.73 <sup>d</sup> ±0.34	0.004 <sup>b</sup> ±0.002	-	-
+LL	4.81 <sup>de</sup> ±0.36	-0.001 <sup>b</sup> ±0.001	-	-
LL ORG	15.66 <sup>ab</sup> ±1.15	0.26 <sup>a</sup> ±0.03	1.02 <sup>bc</sup> ±0.10	0.36 <sup>c</sup> ±0.05
+LL ORG	13.36 <sup>c</sup> ±0.22	0.52 <sup>a</sup> ±0.06	0.88 <sup>c</sup> ±0.05	0.75 <sup>c</sup> ±0.09
eLL	3.12 <sup>e</sup> ±0.55	-0.003 <sup>b</sup> ±0.001	-	-
+eLL	3.39 <sup>e</sup> ±0.76	-0.012 <sup>b</sup> ±0.006	-	-
eLL ORG	15.85 <sup>ab</sup> ±1.89	1.54 <sup>a</sup> ±0.17	1.31 <sup>a</sup> ±0.19	2.20 <sup>a</sup> ±0.25
+eLL ORG	13.97 <sup>bc</sup> ±0.44	1.22 <sup>a</sup> ±0.10	1.09 <sup>abc</sup> ±0.11	1.76 <sup>b</sup> ±0.15
RZ	3.34 <sup>c</sup> ±0.11	0.005 <sup>b</sup> ±0.001	-	-
+RZ	3.01 <sup>c</sup> ±0.25	-0.002 <sup>b</sup> ±0.002	-	-
RZ ORG	16.00 <sup>a</sup> ±0.79	0.49 <sup>a</sup> ±0.17	1.30 <sup>a</sup> ±0.07	0.69 <sup>c</sup> ±0.25
+RZ ORG	15.22 <sup>abc</sup> ±0.54	1.21 <sup>a</sup> ±0.14	1.25 <sup>ab</sup> ±0.04	1.74 <sup>b</sup> ±0.21

eLL soil exhibited slower mineral N turnover than LL and RZ, showing a more gradual decline in NH<sub>4</sub><sup>+</sup>-N and delayed accumulation of NO<sub>3</sub><sup>-</sup>-N accumulation (Fig. 2), which indicates inherently slower N transformation processes. Such clay-rich subsoil material can store organic compounds and mineral N forms like NH<sub>4</sub><sup>+</sup>-N, temporarily protecting them from microbial transformation and plant uptake (Berhe et al. 2018; Castellano et al. 2012; Schoof et al. 2025; Zentgraf et al. 2024). The large surface area and reactive minerals provide sorption sites for NH<sub>4</sub><sup>+</sup>-N and organic N, reducing substrate availability for denitrification (Bingham and Cotrufo 2016; Knicker 2011; Lützow et al. 2006; Rillig et al. 2007; van Groenigen et al. 2015;). Consequently, N becomes less available for microbial processes responsible for N<sub>2</sub>O production (Vogel et al. 2014, 2015), explaining the observed reduction in N<sub>2</sub>O emissions and the lower fertilizer-induced N<sub>2</sub>O-N (Table 2). Nevertheless, the specific mechanism of N retention under digestate application and topsoil dilution remains unclear and requires further investigation. Future studies should identify which soil fractions and clay minerals are involved in N stabilization, determine how long the retained N persists, and assess the extent to which these processes contribute to long-term mitigation of N<sub>2</sub>O emissions. Combining isotopic tracing with fractionation analyses would further clarify how mineral-organic interactions control N stabilization under digestate amendment.

In contrast to the eLL soil, the strongly eroded RZ soil responded to topsoil dilution with an increase in N<sub>2</sub>O emissions (Table 2; Fig. 4d), possibly reflecting the influence of its distinct calcareous properties and relatively higher pH.

As shown in Table S1, RZ already had calcareous conditions in the Ap horizon (pH 7.2; CaCO<sub>3</sub> 3.2%), and mixing 20% subsoil Cc (pH 7.4; CaCO<sub>3</sub> ≈ 11.6%) leaves its pH essentially unchanged (expected ≈ 7.2) while increasing CaCO<sub>3</sub> (≈ 4.9%). Even after dilution, RZ maintained the highest pH compared to LL (≈ 5.7) and eLL (≈ 6.7). Together with the increased CaCO<sub>3</sub> content, this may promote a higher nitrification potential and thus likely contributed to the enhanced N<sub>2</sub>O release observed in this soil (Table S1; Ciu et al. 2012; Gao et al. 2023; Kyveryga et al. 2004; Sahrawat 2008; Tao et al. 2017). Supporting this interpretation, NO<sub>3</sub><sup>-</sup>-N and NH<sub>4</sub><sup>+</sup>-N dynamics revealed rapid NH<sub>4</sub><sup>+</sup>-N depletion between t0 and t1 (Fig. 2c and f), coinciding with an early N<sub>2</sub>O emission peak, which suggests accelerated N turnover and increased nitrification. In contrast, the LL and eLL soils showed slower NH<sub>4</sub><sup>+</sup>-N decline between t2 and t3 (Fig. 2a and b), accompanied by delayed N<sub>2</sub>O peaks (Batlle-Aguilar et al. 2011; Groffman et al. 2002). Thus, in RZ, the combination of calcareous conditions and rapid N turnover promoted greater N<sub>2</sub>O production under dilution, contrasting with the more subdued response observed in LL soils.

In the LL soil, topsoil dilution led to only subtle changes in N<sub>2</sub>O emissions, with no statistically significant increase in cumulative fluxes (Table 2). Nevertheless, fluxes exhibited a slightly prolonged peak (Fig. 4d), corresponding with the observed NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N dynamics, where NH<sub>4</sub><sup>+</sup>-N declined sharply between t1 and t2 and NO<sub>3</sub><sup>-</sup>-N increased thereafter (Fig. 2a and d). This pattern reflects an intermediate timing of N transformation relative to the faster turnover in RZ and the slower processes in eLL. The diluted

LL treatment (+LL ORG) also showed lower CWSN during the N<sub>2</sub>O peak (t1–t2) and again at t3, suggesting enhanced N turnover and reduced retention of mineral N over time (Fig. 1d). This effect likely resulted from mixing C- and N-poor subsoil into the topsoil, which lowered TOC and TN contents (Table 1) and consequently reduced the C: N ratio. Under these conditions, microbial communities may experience an excess of available N relative to C, limiting immobilization and promoting N losses as N<sub>2</sub>O (Akiyama et al. 2020; Baggs et al. 2000; Yao et al. 2022). The minor increase in N<sub>2</sub>O emissions in LL therefore likely reflects a temporary acceleration of N cycling rather than structural differences in soil properties or inherent N availability.

It is important to note that the soils studied (LL, eLL, and RZ) represent distinct erosion states that also differ in parent material and chemistry, such as the carbonate-rich subsoil in RZ (Sommer et al. 2016; Wilken et al. 2020). These inherent differences likely influenced GHG responses alongside erosion effects. Overall, our results show that erosion-induced topsoil dilution alters CO<sub>2</sub> and N<sub>2</sub>O emissions by modifying C and N availability and related microbial processes, underscoring the importance of erosion-aware management when applying organic fertilizers. Integrating such erosion-aware strategies with remote or proximal sensing approaches could support site-specific fertilizer management to improve nutrient efficiency and mitigate GHG emissions.

## 5 Conclusions

This study demonstrated that erosion-induced topsoil dilution affects carbon dioxide (CO<sub>2</sub>) and nitrous oxide (N<sub>2</sub>O) emissions following digestate application differently. Using soil types as proxies for erosion introduces inherent differences in parent material, yet this approach reflects realistic field conditions and provides a robust framework for assessing erosion impacts on carbon (C) and nitrogen (N) cycling. Topsoil dilution consistently reduced CO<sub>2</sub> fluxes, with a significant decline observed only in the non-eroded LL soil, likely due to lower soil organic carbon (SOC) availability and potential C stabilization. In more eroded soils, this effect was weaker, probably because existing C under-saturation limited further mitigation. The response of N<sub>2</sub>O emissions varied with soil properties; reductions occurred in the clay- and aluminium (Al) and iron (Fe) oxide-rich eLL soil, increases in calcareous RZ soil, and only minor increase in LL, likely due to reduced carbon-to-nitrogen ratio (C: N ratio). These results highlight that erosion state and soil properties strongly influence greenhouse gas (GHG) responses to digestate. Future research should test these processes under field conditions across different management and climatic settings and develop site-specific fertilization

strategies tailored to local soil and erosion characteristics. In erosion-affected landscapes with high spatial variability, remote or proximal sensing can help identify key soil differences such as pH, texture, and organic matter content to adjust fertilizer application, improve nutrient efficiency, and minimize N<sub>2</sub>O emissions.

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**Data Availability** The datasets generated during and/or analysed during the current study are available in the BonaRes repository, [<https://doi.org/10.4228/zalf-76tz-3k46>]

## Declarations

**Ethical approval** This manuscript has not been published elsewhere and is not under consideration by any other journal. Furthermore, there are no conflicts with the ethical responsibilities formulated by Springer (<https://www.springer.com/journal/11104/submission-guidelines#Instructions> for authors\_Ethical Responsibilities of Authors).

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